# Polymer Chemistry

### PAPER



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### Introduction

Supramolecular polymerization is now widely recognized as a strategy tool for the creation of polymeric materials.<sup>1-4</sup> In contrast to classical polymers, the presence of potentially reversible non-covalent interactions provides access to outstanding functionalities such as stimuli-responsive or self-healing characteristics. Through manipulating a range of non-covalent intermolecular interactions, such as hydrogen-bonding,  $\pi$ - $\pi$  stacking, hydrophobic–hydrophobic and/or electrostatic interactions, it is possible to form linear supramolecular polymers rather than undefined aggregates to fulfill a multitude of tasks. In particular,  $\pi$ -conjugated molecules and polymers that form stacked structures have attracted much attention as

## Controlling the supramolecular polymerization of dinuclear isocyanide gold( $\iota$ ) arylethynylene complexes through tuning the central $\pi$ -conjugated moiety<sup>+</sup>

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Three dinuclear gold(i) *p*-arylene(diethynyl) complexes with the same peripheral isocyanide ligation and different central conjugated moieties have been prepared and their supramolecular polymerization was comprehensively studied. The central, core-forming  $\pi$ -conjugated arylene moieties present in the complexes have been shown to exhibit a pronounced influence on the resulting photophysical properties, self-assembly mechanisms, morphologies and gelation behavior of the supramolecular polymers. Complexes **1** and **2** bearing biphenylene and phenylene central chromophores exhibit a cooperative mechanism during the supramolecular polymerization, with the appearance of pronounced low-energy phosphorescence attributed to the aurophilic interactions. In contrast, compound **3** with a central benzothiadiazole moiety was found to undergo self-assembly *via* an isodesmic mechanism. Due to the presence of stronger  $\pi$ - $\pi$  interactions, complexes **1** and **3** possess lower  $\Delta G_0$  values for self-assembly and a stronger tendency to form metallogels. Given the balance between the non-covalent interaction enthalpy of the core-forming moieties and intercoronal chain repulsion, **1** and **3** tend to form one-dimensional fibers, while **2** with a short central chromophore is likely to form zero-dimensional spherical aggregates.

building blocks due to their wide range of applications which include light-harvesting, sensing, and photocatalysis.5,6 The optoelectronic properties of these aggregates strongly depend on the  $\pi$ -conjugated central core and their intermolecular orientations. A variety of  $\pi$  systems, such as polythiophene,<sup>7,8</sup> polyfluorene,<sup>9,10</sup> perylene diimide,<sup>11</sup> oligo(*p*-phenyleneviny-lene),<sup>12</sup> azobenzene,<sup>13</sup> porphyrin,<sup>14,15</sup> hexabenzocoronenes,<sup>16</sup> and Pt(II) metal complexes,<sup>17–24</sup> have been successfully utilized to fabricate supramolecular polymers for various applications. In many cases, strong  $\pi$ - $\pi$  stacking of the chromophore leads to significant emission quenching, which is undesirable for utilizing their luminescence properties. In the last decade, mechanistic insights into supramolecular polymerization have been significantly enriched.<sup>1–3,25</sup> Two different types of mechanisms, namely isodesmic and cooperative processes, have been demonstrated to be involved in most of the self-assembly systems.<sup>1-3</sup> Recent advances have witnessed the seeded living self-assembly of  $\pi$  conjugated organic molecules to provide access to complex nanoscale 1D and 2D morphologies with unprecedented levels of control of dimensions via cooperative mechanisms.<sup>7-21</sup> Minor structural variations have been demonstrated to exhibit a significant effect on self-assembly mechanisms.

The chemistry of organogold(i) complexes has received much attention because of the presence of aurophilic Au(i)-Au(i)

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# 证 明

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